

# JTH01/ABINIT

JTH1 PAW dataset / ABINIT 7.5.3

name and version of the code: ABINIT 7.5.3  
type of basis set: plane waves  
method: projector-augmented wave (JTH1)

## GENERAL INFORMATION

exchange-correlation functional	PBE
relativistic scheme	core and valence scalar relativistic (Koelling-Harmon)
assignment of core / valence states	see table
basis set size	cutoff energy = 20 Ha
k-mesh density	6 750/ $N$ k-points in the full first Brillouin zone of an $N$ -atom cell
reciprocal-space integration method	Fermi-Dirac smearing with a fictitious temperature corresponding to 0.002 Ha

## METHOD-SPECIFIC INFORMATION

PAW radii	see table ( $R_{PAW}$ )
PAW cutoff energy	20 Ha
PAW cutoff augmentation energy	40 Ha
Partial-wave basis size	2 elements per $\ell$ -value except for H, He, Li (one p element) and Ga, Ge, In, K, Kr, Pb and Sn (one d element)

## ADDITIONAL COMMENTS

PAW dataset generator	ATOMPAW 3.1.0.2
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## REFERENCES

PAW dataset

- [1] <http://pwpaw.wfu.edu>
- [2] <http://www.abinit.org/downloads/PAW2>
- [3] F. Jollet, M. Torrent and N. Holzwarth, *Comput. Phys. Commun.* **185**, 1246–1254 (2014).

code

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- [6] M. Torrent, F. Jollet, F. Bottin, G. Zerah, X. Gonze, *Comput. Mater. Sci.* **42**, 337–351, (2008).

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- [7] D. D. Koelling and B. N. Harmon, *J. Phys. C: Solid State* **10**, 3107–3114 (1977).

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**Table I.** Calculation settings and results per element: valence, PAW radius  $R_{PAW}$ , equilibrium volume per atom  $V_0$ , bulk modulus  $B_0$ , pressure derivative of the bulk modulus  $B_1$ .

	valence	$R_{PAW}$ [a.u.]	$V_0$ [ $\text{\AA}^3/\text{atom}$ ]	$B_0$ [GPa]	$B_1$ [-]
H	1s <sup>1</sup> 2p <sup>0</sup>	1.0	17.42891	10.28346	3.719
He	1s <sup>2</sup> 2p <sup>0</sup>	1.3	17.82441	0.83567	6.413
Li	1s <sup>2</sup> 2s <sup>1</sup> 2p <sup>0</sup>	1.6	20.23673	13.80591	3.344
Be	1s <sup>2</sup> 2s <sup>2</sup>	1.3	7.90902	122.23427	2.891
B	2s <sup>2</sup> 2p <sup>1</sup>	1.7	7.24585	237.21331	3.487
C	2s <sup>2</sup> 2p <sup>2</sup>	1.55	11.66043	213.85783	2.594
N	2s <sup>2</sup> 2p <sup>3</sup>	1.2	28.93696	55.05401	2.963
O	2s <sup>2</sup> 2p <sup>4</sup>	1.4	18.58836	51.34026	3.880
F	2s <sup>2</sup> 2p <sup>5</sup>	1.4	19.29223	35.23528	4.276
Ne	2s <sup>2</sup> 2p <sup>6</sup>	1.8	24.23408	1.33457	13.286
Na	2s <sup>2</sup> 2p <sup>6</sup> 3s <sup>1</sup>	1.8	37.11691	7.99425	3.227
Mg	2s <sup>2</sup> 2p <sup>6</sup> 3s <sup>2</sup>	1.9	22.99185	35.65583	4.271
Al	3s <sup>2</sup> 3p <sup>1</sup>	1.9	16.47564	77.73267	4.560
Si	3s <sup>2</sup> 3p <sup>2</sup>	1.9	20.53424	88.13173	4.364
P	3s <sup>2</sup> 3p <sup>3</sup>	1.9	21.57113	67.40838	4.364
S	3s <sup>2</sup> 3p <sup>4</sup>	1.9	17.31878	80.58862	3.863
Cl	3s <sup>2</sup> 3p <sup>5</sup>	1.8	39.16074	18.46534	4.448
Ar	3s <sup>2</sup> 3p <sup>6</sup>	1.8	52.23071	0.74955	6.567
K	3s <sup>2</sup> 3p <sup>6</sup> 4s <sup>1</sup> 3d <sup>0</sup>	2.1	73.77139	3.62522	3.658
Ca	3s <sup>2</sup> 3p <sup>6</sup> 4s <sup>2</sup> 3d <sup>0</sup>	1.9	42.13974	17.44360	3.823
Sc	3s <sup>2</sup> 3p <sup>6</sup> 4s <sup>2</sup> 3d <sup>1</sup>	2.4	24.64343	54.20351	3.356
Ti	3s <sup>2</sup> 3p <sup>6</sup> 4s <sup>1</sup> 3d <sup>3</sup>	2.3	17.43943	111.33248	3.582
V	3s <sup>2</sup> 3p <sup>6</sup> 4s <sup>2</sup> 3d <sup>3</sup>	2.2	13.50749	180.77544	3.818
Cr	3s <sup>2</sup> 3p <sup>6</sup> 4s <sup>1</sup> 3d <sup>5</sup>	2.1	11.89846	174.61933	6.518
Mn	3s <sup>2</sup> 3p <sup>6</sup> 4s <sup>1</sup> 3d <sup>6</sup>	2.1	11.58353	110.57780	2.073
Fe	3s <sup>2</sup> 3p <sup>6</sup> 4s <sup>1</sup> 3d <sup>7</sup>	2.1	11.46507	185.27502	4.325
Co	3s <sup>2</sup> 3p <sup>6</sup> 4s <sup>1</sup> 3d <sup>8</sup>	2.1	10.94555	209.20659	5.147
Ni	3s <sup>2</sup> 3p <sup>6</sup> 4s <sup>2</sup> 3d <sup>8</sup>	2.1	10.98538	195.04959	4.806
Cu	3s <sup>2</sup> 3p <sup>6</sup> 4s <sup>1</sup> 3d <sup>10</sup>	2.2	12.00941	139.35753	5.169
Zn	4s <sup>2</sup> 4p <sup>0</sup> 3d <sup>10</sup>	2.3	15.31269	74.63335	5.868
Ga	4s <sup>2</sup> 4p <sup>1</sup> 3d <sup>10</sup>	2.31	20.39324	49.90232	5.286
Ge	4s <sup>2</sup> 4p <sup>2</sup> 3d <sup>10</sup>	2.3	24.16805	59.14874	4.895
As	4s <sup>2</sup> 4p <sup>3</sup>	2.2	22.55430	68.80630	4.400
Se	4s <sup>2</sup> 4p <sup>4</sup>	2.2	29.96519	46.92162	4.420
Br	4s <sup>2</sup> 4p <sup>5</sup>	2.2	39.76958	22.16956	4.958
Kr	4s <sup>2</sup> 4p <sup>6</sup> 4d <sup>0</sup>	2.2	65.61745	0.66575	7.041
Rb	4s <sup>2</sup> 4p <sup>6</sup> 5s <sup>1</sup> 4d <sup>0</sup>	2.3	91.11045	2.75660	3.576
Sr	4s <sup>2</sup> 4p <sup>6</sup> 5s <sup>1</sup> 4d <sup>1</sup>	2.21	54.63677	11.62676	3.449
Y	4s <sup>2</sup> 4p <sup>6</sup> 5s <sup>2</sup> 4d <sup>1</sup>	2.21	32.88362	40.84671	3.093
Zr	4s <sup>2</sup> 4p <sup>6</sup> 5s <sup>1</sup> 4d <sup>3</sup>	2.21	23.39812	93.38537	3.308
Nb	4s <sup>2</sup> 4p <sup>6</sup> 5s <sup>1</sup> 4d <sup>4</sup>	2.21	18.13594	169.36451	3.672
Mo	4s <sup>2</sup> 4p <sup>6</sup> 5s <sup>1</sup> 4d <sup>5</sup>	2.2	15.81129	259.30147	4.058
Tc	4s <sup>2</sup> 4p <sup>6</sup> 5s <sup>1</sup> 4d <sup>6</sup>	2.2	14.46649	301.41262	4.603
Ru	4s <sup>2</sup> 4p <sup>6</sup> 5s <sup>1</sup> 4d <sup>7</sup>	2.2	13.82013	307.96378	4.297
Rh	4s <sup>2</sup> 4p <sup>6</sup> 5s <sup>1</sup> 4d <sup>8</sup>	2.4	14.08022	261.48262	5.066
Pd	4s <sup>2</sup> 4p <sup>6</sup> 5s <sup>1</sup> 4d <sup>9</sup>	2.5	15.32170	166.63735	5.830
Ag	5s <sup>1</sup> 5p <sup>0</sup> 4d <sup>10</sup>	2.5	17.82260	91.32391	5.846
Cd	5s <sup>2</sup> 5p <sup>0</sup> 4d <sup>9.5</sup>	2.5	22.76797	44.17371	6.759
In	5s <sup>2</sup> 5p <sup>1</sup> 5d <sup>0</sup>	2.9	27.56282	36.88450	5.224

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Sn	$5s^2 5p^2 5d^0$	2.7	36.90882	36.05080	4.883
Sb	$5s^2 5p^3$	2.3	31.81352	50.68865	4.654
Te	$5s^2 5p^4$	2.3	34.97375	45.13662	4.735
I	$5s^2 5p^5$	2.3	50.41409	18.72654	5.018
Xe	$5s^2 5p^6$	2.4	87.14873	0.53582	4.918
Cs	$5s^2 5p^6 6s^1 5d^0$	2.45	117.82868	2.05877	1.843
Ba	$5s^2 5p^6 6s^2 5d^0$	2.3	63.23845	8.77715	2.243
Lu	$5s^2 5p^6 6s^2 5d^1 4f^{14}$	2.5	29.05223	47.68307	3.519
Hf	$5s^2 5p^6 6s^2 5d^2$	2.41	22.54013	107.62002	3.303
Ta	$5s^2 5p^6 6s^2 5d^3$	2.41	18.30301	195.94631	3.794
W	$5s^2 5p^6 6s^2 5d^4$	2.41	16.17507	303.98275	4.152
Re	$5s^2 5p^6 6s^2 5d^5$	2.4	14.98499	364.96449	4.376
Os	$5p^6 6s^1 5d^7$	2.5	14.32062	425.11571	4.549
Ir	$5p^6 6s^1 5d^8$	2.5	14.55237	347.21064	5.034
Pt	$6s^1 6p^0 5d^9$	2.5	15.68200	250.05791	5.560
Au	$6s^1 6p^0 5d^{10}$	2.5	18.01243	140.30702	5.501
Hg	$6s^2 6p^0 5d^{10}$	2.4	29.65450	8.84950	12.145
Tl	$6s^2 6p^1$	2.9	31.42413	27.78389	5.431
Pb	$6s^2 6p^2 6d^0$	2.9	32.02008	41.11044	5.147
Bi	$6s^2 6p^3$	2.9	36.92694	43.22333	4.717
Po	$6s^2 6p^4$	2.9	37.57228	46.07731	4.848
Rn	$6s^2 6p^6$	2.3	92.80198	0.61546	8.807